

Write Titrating Volume (Spreads: 1.00)

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Acetic Acid (pKa = 4.75) titrated with NaOH

a. 0.100 M / volume 10.00 mL / pH 3.75 / 0.100 M

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c. 0.100 M / pH 4.75 / compound that releases acetate
0.100 M / 10.00 mL / 0.100 M / 0.100 M

Acetic Acid (pKa = 4.75) titrated with NaOH

0.100 M / 10.00 mL / 0.100 M / 0.100 M
0.100 M / 10.00 mL / 0.100 M / 0.100 M

Species	Acid (pH, Base pH, or Salt pH)	Acids with M_1 ($M_1 = 0.10$)	Conductivity (κ or Λ)
HCl	H^+	H^+	H^+
HCl	H^+	H^+	H^+
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