

Element: Pure Type of Matter: Substance
One smallest parts
 - represented by atom or ion symbol
 - example: Helium He

Compound: Two (2) or more elements chemically
combined
 - represented by chemical formula
 - example: water H₂O
 - e.g. oxygen O₂, hydrogen H₂ and
nitrogen N₂
 - combined in fixed proportion

Mixture: Two or more substances physically
combined
 - do not combine in some proportion
 - heterogeneous mixture - e.g. oil & water
 - homogeneous mixture (solution) - alcohol or
salt solution or even brine
 - have variable composition

Note Taking Guide Episode 150name

Brad Jones



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