

## Chapter 6 Chemical Bonds

**Section 6.2 Covalent Bonding****(pages 165–169)**

*This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar. It also discusses attractions between molecules.*

**Reading Strategy (page 165)**

**Relating Text and Visuals** As you read the section, look closely at Figure 9. Complete the table by describing each type of model shown. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Molecular Models	
Model	Description
Electron dot	
Structural formula	
Space-filling	
Electron cloud	

**Covalent Bonds (pages 165–167)**

- Describe a covalent bond. \_\_\_\_\_  
\_\_\_\_\_
- Circle the letters of molecular models that show orbitals of atoms overlapping when a covalent bond forms.
  - electron dot
  - structural formula
  - space-filling
  - electron cloud
- Describe a molecule. \_\_\_\_\_  
\_\_\_\_\_
- Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the protons in each nucleus. \_\_\_\_\_
- Circle the correct answer. Nitrogen has five valence electrons. How many pairs of electrons must two nitrogen atoms share in order for each atom to have eight valence electrons?
  - zero
  - one
  - two
  - three

# Note Taking Guide Covalent Bonding Answer Key

**SA Adler**



## Note Taking Guide Covalent Bonding Answer Key:

*Reading and Writing in Science* Maria C. Grant, Douglas Fisher, Diane Lapp, 2015-01-21 Engage your students in scientific thinking across disciplines Did you know that scientists spend more than half of their time reading and writing Students who are science literate can analyze present and defend data both orally and in writing The updated edition of this bestseller offers strategies to link the new science standards with literacy expectations and specific ideas you can put to work right away Features include A discussion of how to use science to develop essential 21st century skills Instructional routines that help students become better writers Useful strategies for using complex scientific texts in the classroom Tools to monitor student progress through formative assessment When students are curious they thrive Give your students the strong base they need to create and share scientific ideas that have an impact in the classroom and beyond This is a teacher friendly book that drew me in from the introduction to the end Through real life scenarios combined with useful methods for instruction it illustrates how science teachers can use language as a tool for teaching science Trina Allen Science Content Specialist Measurement Incorporated An eminently readable guide for the novice and experienced teacher The many practical ideas in this volume demonstrate that improving students skills in reading and writing can also improve their understanding and ability in science Cary Sneider Associate Research Professor Portland State University Portland OR *Focus on Physical Science California Edition* Michael J. Padilla, 2007 [O Level Chemistry Questions and Answers PDF](#) Arshad Iqbal, The O Level Chemistry Quiz Questions and Answers PDF IGCSE GCSE Chemistry Competitive Exam Questions Chapter 1 14 Practice Tests Class 9 10 Chemistry Textbook Questions for Beginners includes revision guide for problem solving with hundreds of solved questions O Level Chemistry Questions and Answers PDF book covers basic concepts analytical and practical assessment tests O Level Chemistry Quiz PDF book helps to practice test questions from exam prep notes The O Level Chemistry Quiz Questions and Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved tests O Level Chemistry Questions and Answers PDF Free download chapter 1 a book covers solved common questions and answers on chapters Acids and bases chemical bonding and structure chemical formulae and equations electricity electricity and chemicals elements compounds mixtures energy from chemicals experimental chemistry methods of purification particles of matter redox reactions salts and identification of ions and gases speed of reaction and structure of atom tests for school and college revision guide Chemistry Interview Questions and Answers PDF Download free eBook sample covers beginner s solved questions textbook s study notes to practice online tests The IGCSE GCSE Chemistry Interview Questions Chapter 1 14 PDF book includes high school question papers to review practice tests for exams O Level Chemistry Practice Tests a textbook s revision guide with chapters tests for IGCSE NEET MCAT GRE GMAT SAT ACT competitive exam O Level Chemistry Questions Bank Chapter 1 14 PDF book covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Acids and Bases Questions Chapter 2 Chemical Bonding

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*Chemistry Extension File* Eileen Ramsden, 2002 This chemistry extension file includes teaching notes guidance on coursework activities and equipment It has at least one assignment for each topic in the textbooks suitable for classwork and homework A comprehensive range of practical activities are included It contains extensive Key Skills and ICT materials An exam file resource containing a complete set of exam style questions in a format that can be used throughout Years 10 and 11 or as a resource for a revision programme is included

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*Study Guide for Chemical Principles* Thomas Elliott Taylor, 1979      *The Pearson Guide To The Scra Examination, 2/E* Thorpe, 2008-09      **UPSC Prelims Exam**

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