natural transmutation - Uranium spontaneously decays.

$$^{238}_{92}U \rightarrow ^{234}_{90}Th + ^{4}_{2}He$$

<u>artificial transmutation</u> - bombardment of a stable isotope to force it to decay.

$$^{14}_{7}N + ^{4}_{2}He \rightarrow ^{1}_{1}p + ^{17}_{6}D$$

When the bullets are + charged, they are repelled by the nucleus they are bombarding. To overcome the repulsions, they must be outstand to very high speeds by Posticia

nuclear fission - Heavy nuclei are bombarded with neutrons and split. 1 1 ^{235}II $> ^{142}D$ = 1 ^{91}V = 1 = 2 1 = 1

$$_{0}^{1}n + _{92}^{235}U \rightarrow _{56}^{142}Ba + _{36}^{91}Kr + 3(_{0}^{1}n) + E$$

Mass of particles produced is slightly <u>more</u> than the mass of the reactants.

This mass is converted into <u>energy</u>. (E = m c²)

required for a Chain sealtion

nuclear reactors: control fission <u>Chain</u> reactions to produce energy dangers:

nuclear fusion - combination of 2 nuclei into 1 with release of

$${}_{1}^{2}H + {}_{1}^{3}H \rightarrow {}_{2}^{4}He + {}_{0}^{1}n + E$$

Mass of particles produced is much MOTE than the mass of the TEXTENS.

This MOSS is converted into energy. (E = MC)

On back, list advantages of and problems with using fusion as an energy source.

CHEMISTRY: A Study of Matter

The issues with using fusion as energy generally just come from concern for the public and their safety due to storage of nuclear material for a long time causing concerns well as radioactivity being a big issue.

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Table of Contents Note Taking Guide 1502 Chemistry

- 1. Understanding the eBook Note Taking Guide 1502 Chemistry
 - The Rise of Digital Reading Note Taking Guide 1502 Chemistry
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Note Taking Guide 1502 Chemistry
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Note Taking Guide 1502 Chemistry
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Note Taking Guide 1502 Chemistry
 - Personalized Recommendations
 - Note Taking Guide 1502 Chemistry User Reviews and Ratings
 - Note Taking Guide 1502 Chemistry and Bestseller Lists
- 5. Accessing Note Taking Guide 1502 Chemistry Free and Paid eBooks

- Note Taking Guide 1502 Chemistry Public Domain eBooks
- Note Taking Guide 1502 Chemistry eBook Subscription Services
- Note Taking Guide 1502 Chemistry Budget-Friendly Options
- 6. Navigating Note Taking Guide 1502 Chemistry eBook Formats
 - o ePub, PDF, MOBI, and More
 - Note Taking Guide 1502 Chemistry Compatibility with Devices
 - Note Taking Guide 1502 Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Note Taking Guide 1502 Chemistry
 - Highlighting and Note-Taking Note Taking Guide 1502 Chemistry
 - Interactive Elements Note Taking Guide 1502 Chemistry
- 8. Staying Engaged with Note Taking Guide 1502 Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Note Taking Guide 1502 Chemistry
- 9. Balancing eBooks and Physical Books Note Taking Guide 1502 Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Note Taking Guide 1502 Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Note Taking Guide 1502 Chemistry
 - Setting Reading Goals Note Taking Guide 1502 Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Note Taking Guide 1502 Chemistry
 - Fact-Checking eBook Content of Note Taking Guide 1502 Chemistry
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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