

Reaction Energy

Reaction Quiz: Thermodynamics

In the space provided, write the letter of the best answer for each multiple-choice question or the appropriate calculation.

1. The sign convention for the study of thermodynamics is that
 - a. energy is always being created.
 - b. energy is always being destroyed.
 - c. energy is being destroyed.
 - d. energy is conserved.
2. Energy is measured in units of
 - a. joules.
 - b. calories.
 - c. grams.
 - d. pounds.
3. Heat gained is used to calculate enthalpy
 - a. ΔH .
 - b. ΔG .
 - c. ΔS .
 - d. None of these.
4. The specific heat of a substance is the amount of energy needed to
 - a. 1 g of the substance.
 - b. 1 mole of the substance.
 - c. 1 g of the substance.
 - d. 1 mole of the substance.
5. Methods of the following are the amount of enthalpy (heat energy) required to
 - a. increase the temperature of a substance.
 - b. the specific heat of the substance.
 - c. the mass of the substance.
 - d. the temperature change.

Reaction Energy Section Quiz Modern Chemistry

N Colangelo



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