






3-*f* 4-*c* 4-*m* 4-*d*

b) Dibromide bis(ethane-1,2-diamine) Cobalt(III) is the IUPAC name of $[\text{CoBr}_2(\text{en})_2]^+$ complex.

$[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ in solution is three



11 3 3 3 3

- 012

④ Zeta Potential

(II) Why a negatively charged sol is obtained when AgNO_3 solution is added to KI solution?*

131

Zn, Cd, Hg are not regarded as transition metals because they have completely filled d -subshell with outer electronic configuration $(n-1)d^{10}ns^2$

9. (a) What happens when:

11-3-31

(ii) Ethanal is treated with excess ethanol and acid?

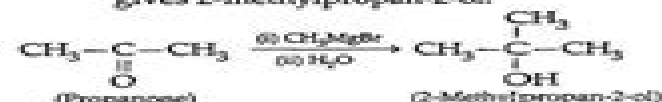
(iii) Methanal undergoes Cannizzaro reaction?

DE

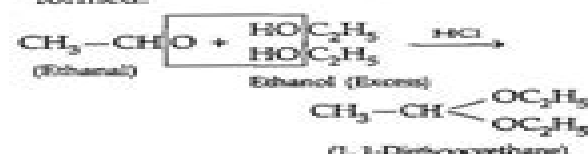
(b) Write the main product in the following reactions: [1 × 3 = 3]



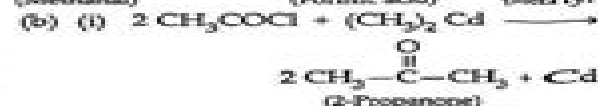
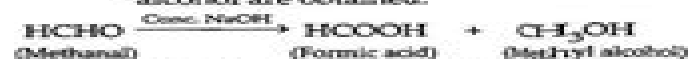
Ans. (a) (i) When propanone is treated with CH_3MgBr followed by hydrolysis it gives 2-methylpropan-2-ol.



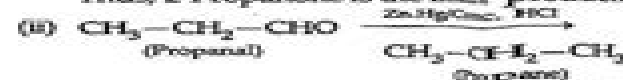
(ii) When ethanal reacts with excess of ethanol in presence of trace amount of HCl then, 1, 1-Dithioxyethane will be formed.



(iii) When methanal undergoes Cannizzaro reaction then, formic acid and methyl alcohol are obtained.



Thus, 2-Propanone is the main product.



Thus, propane is the main product.



Thus, cyclohexane is the main product.

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