

- **Positively charged:** electrons are removed making the object electron **deficient**.
- **Negatively charged:** electrons are added giving the object an **excess** of electrons.

So in practise what happens is that the number of positive charges (protons) remains the same and the number of electrons changes:

There are:

6 positive charges and
6 negative charges

$$6 + (-6) = 0$$



There is zero net charge:
The object is neutral

6 positive charges and
4 negative charges

$$6 + (-4) = +2$$



The net charge is +2
The object is positively charged

6 positive charges and
9 negative charges

$$6 + (-9) = -3$$



The net charge is -3
The object is negatively charged

Tribo-electric charging



Tip

Charge, like energy, cannot be created or destroyed. We say that charge is **conserved**.

Objects may become charged in many ways, including by contact with or being rubbed by other objects. This means that they can gain or lose negative charge. For example, charging happens when you rub your feet against the carpet. When you then touch something metallic or another person, you feel a shock as the excess charge that you have collected is discharged.

When you rub your feet against the carpet, negative charge is transferred to you from the carpet. The carpet will then become positively charged by the same amount.

📺 See video: VPfns at www.everythingscience.co.za

Another example is to take two *neutral* objects such as a plastic ruler and a cotton cloth (handkerchief). To begin, the two objects are neutral (i.e. have the same amounts of positive and negative charge).

Note: We represent the positive charge with a + and the negative charge with a -. This is just to **illustrate the balance and changes that occur, not the actual location** of the positive and negative charges. The charges are spread throughout the material and the real change happens by increasing or decreasing electrons on the surface of the materials.

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Raffaella Di Napoli



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Table of Contents Physical Science Grade11 Term3 Electrostatics

1. Understanding the eBook Physical Science Grade11 Term3 Electrostatics
 - The Rise of Digital Reading Physical Science Grade11 Term3 Electrostatics
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Grade11 Term3 Electrostatics
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Grade11 Term3 Electrostatics
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Grade11 Term3 Electrostatics

- Personalized Recommendations
- Physical Science Grade11 Term3 Electrostatics User Reviews and Ratings
- Physical Science Grade11 Term3 Electrostatics and Bestseller Lists
- 5. Accessing Physical Science Grade11 Term3 Electrostatics Free and Paid eBooks
 - Physical Science Grade11 Term3 Electrostatics Public Domain eBooks
 - Physical Science Grade11 Term3 Electrostatics eBook Subscription Services
 - Physical Science Grade11 Term3 Electrostatics Budget-Friendly Options
- 6. Navigating Physical Science Grade11 Term3 Electrostatics eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Grade11 Term3 Electrostatics Compatibility with Devices
 - Physical Science Grade11 Term3 Electrostatics Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Grade11 Term3 Electrostatics
 - Highlighting and Note-Taking Physical Science Grade11 Term3 Electrostatics
 - Interactive Elements Physical Science Grade11 Term3 Electrostatics
- 8. Staying Engaged with Physical Science Grade11 Term3 Electrostatics
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Grade11 Term3 Electrostatics
- 9. Balancing eBooks and Physical Books Physical Science Grade11 Term3 Electrostatics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Grade11 Term3 Electrostatics
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Grade11 Term3 Electrostatics
 - Setting Reading Goals Physical Science Grade11 Term3 Electrostatics
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Grade11 Term3 Electrostatics

- Fact-Checking eBook Content of Physical Science Grade11 Term3 Electrostatics
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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