

Physical Science Chapter 4

1000 points

Thermography

Exotherm

Endotherm

Heat

Freeze

Boiling

Melting

Condensing

Evaporation/boiling point

Specific Heat

The specific heat of soil is $0.20 \text{ kcal/kg}^\circ\text{C}$ and the specific heat of water is $1.00 \text{ kcal/kg}^\circ\text{C}$. This means that if 1 kg of soil and 1 kg of water each receives 1 kcal of energy, ideally,

The soil will be 5°C warmer than the water.

The specific heat of water is $1.00 \text{ cal/g}^\circ\text{C}$ and the specific heat of air is $0.167 \text{ cal/g}^\circ\text{C}$. The same amount of energy applied to equal

Physical Science Chapter 4 Test

Caren Van Slyke



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