

QUESTION 2

- 2.1 The graph of molecular size versus the boiling point is given below.
The letters **A**, **B** and **C** represent the compounds CH_4 , C_2H_6 and C_3H_8 respectively.



- 2.1.1 Define the term *boiling point*. (2)
- 2.1.2. Describe the trend in the boiling points of the compounds as shown by the graph. (2)
- 2.1.3. Explain the answer to QUESTION 2.1.2 by referring to MOLECULAR SIZE, TYPE and STRENGTH of INTERMOLECULAR FORCES. (3)
- 2.1.4 Which ONE of the compounds (**A**, **B** or **C**) has the HIGHEST vapour pressure? Explain the answer by referring to the data on the graph. (2)
- 2.2 Consider the two molecules in the table below.

NAME OF SUBSTANCE	FORMULA	MOLACULAR MASS (g)	BOILING POINT ($^{\circ}\text{C}$)
Ammonia	NH_3	17	- 33
Phosphine	PH_3	x	- 87,4

- 2.2.1 Determine the molecular mass (x) of phosphine (1)
- 2.2.2 Explain the difference in the boiling points by referring to the TYPE and STRENGTH of INTERMOLECULAR FORCES. (3)

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Lei Huang



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can be predicted a very careful and detailed effort is needed to generate sunspot numbers This book deals with this topic together with several others that present related phenomena that all indicate the physical processes that take place in the Sun and its exterior environment The reviews in the book also present the latest theoretical and modelling studies that attempt to explain the activity cycle It remains true as has been shown in the unexpected characteristics of the first two solar cycles in the 21st century that predictability remains a serious challenge Nevertheless the highly expert and detailed reviews in this book using the very best solar observations from both ground and space based telescopes provide the best possible report on what is known and what is yet to be discovered Originally published in Space Science Reviews Vol 186 Issues 1 4 2014

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