

temperature:

- a measure of the average kinetic energy of the particles in a sample of matter
- does not depend on the amount of matter in the sample
- symbol is T; unit is °C

heat:

- total amount of kinetic energy that flows because of a difference in temperature
- depends on mass of sample
- symbol is Q; unit is J (1 J = 4.18 cal)

Kinetic energy is energy in motion

Potential energy is store

- Potential energy is hiding and cannot be measured
- Only changes in P.E. can be measured.

specific heat capacity:

- amount of energy required to raise the temperature of 1 gram of substance 1 °C
- symbol is c; unit is J/g°C

$$\begin{array}{c}
 \text{mass} \swarrow \\
 Q = m \times C \times \Delta t \\
 \swarrow \quad \quad \quad \nwarrow \\
 \text{heat} \quad \quad \quad \text{specific heat} \quad \quad \quad \text{change in temperature}
 \end{array}$$

When heat (Q) is absorbed by a system, part of it (C) goes into storage as potential energy and part of it is used to make the molecules move around more, raising the temperature (Δt).

Why does sand get hotter in the day and colder at night than the water?

Note Taking Guide Episode 6chemical Formula Key

Ying Liu



Note Taking Guide Episode 6chemical Formula Key:

This book delves into Note Taking Guide Episode 6chemical Formula Key. Note Taking Guide Episode 6chemical Formula Key is a vital topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Note Taking Guide Episode 6chemical Formula Key, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Note Taking Guide Episode 6chemical Formula Key
 - Chapter 2: Essential Elements of Note Taking Guide Episode 6chemical Formula Key
 - Chapter 3: Note Taking Guide Episode 6chemical Formula Key in Everyday Life
 - Chapter 4: Note Taking Guide Episode 6chemical Formula Key in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, this book will provide an overview of Note Taking Guide Episode 6chemical Formula Key. This chapter will explore what Note Taking Guide Episode 6chemical Formula Key is, why Note Taking Guide Episode 6chemical Formula Key is vital, and how to effectively learn about Note Taking Guide Episode 6chemical Formula Key.
 3. In chapter 2, this book will delve into the foundational concepts of Note Taking Guide Episode 6chemical Formula Key. This chapter will elucidate the essential principles that need to be understood to grasp Note Taking Guide Episode 6chemical Formula Key in its entirety.
 4. In chapter 3, this book will examine the practical applications of Note Taking Guide Episode 6chemical Formula Key in daily life. The third chapter will showcase real-world examples of how Note Taking Guide Episode 6chemical Formula Key can be effectively utilized in everyday scenarios.
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 6. In chapter 5, this book will draw a conclusion about Note Taking Guide Episode 6chemical Formula Key. The final chapter will summarize the key points that have been discussed throughout the book.
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