

### Atomic Number

- Increase across a period
- increase down a family

### Atomic Mass

- increase across a period
- increase down a family

### Atomic Radius

half the distance between nuclei of two like atoms.

- decreases across a period (due to increased attraction of more positive nuclei to electron clouds)
- increase down a family (due to addition of an energy level)

### Electron Affinity

amount of energy when an atom gains an electron

- increases across a period
- decreases down a family

### Ionization Energy

amount of energy required to remove an electron from a neutral atom

- increase across a period
- decrease down a family

### Electron Distribution - Forming Ions

- Atoms become ions by gaining or losing electrons to become an ion.
- An atom is stable when it has a full outer energy level (8 valence electrons).
- Atoms that lose (give away) electrons become positive charged ions. Atoms that gain (steal) electrons become negative charged ions.

# Note Taking Guide Chemistry Episode 403

**CL Gary**



## **Note Taking Guide Chemistry Episode 403:**

## Reviewing **Note Taking Guide Chemistry Episode 403**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Note Taking Guide Chemistry Episode 403**," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book's central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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