### **Equilibrium Practice Problems**

1. Write the equilibrium expression for each of the following reactions:

$$N_{2 (g)} + 3 H_{2 (g)} \leftrightarrow 2 NH_{3 (g)}$$

$$K = \frac{[NH_3]^2}{[N_2][H_2]^3}$$

$$I_{2(x)} + CI_{2(g)} \leftrightarrow 2ICI_{(g)}$$

$$K = \underline{\prod_{[Cl_2]}}$$

$$NO_{2(g)} \leftrightarrow NO_{(g)} + 1/2 O_{2(g)}$$

$$K = [NO][O_2]^{1/2}$$
  
 $[NO_2]$ 

 The dissociation of acetic acid, CH<sub>3</sub>COOH, has an equilibrium constant at 25°C of 1.8 x 10<sup>-5</sup>. The reaction is

If the equilibrium concentration of  $CH_3COOH$  is 0.46 moles in 0.500 L of water and that of  $CH_3COO'$  is 8.1 x  $10^{-3}$  moles in the same 0.500 L, calculate [H $^+$ ] for the reaction.

$$K = [CH_3COO^*][H^*]$$

$$[CH_3COOH]$$

Plug in known values and solve.

$$1.8 \times 10^{-5} = [8.1 \times 10^{-3} \text{ moles/0.500 L}] [H^*]$$
  
[0.46 moles/0.500 L]

$$[H^+] = 1.0 \times 10^{-3} M$$

Indicate the effect of a catalyst, pressure, temperature and concentration on each of the following on:

		catalyst	pressure	temperature	concentration
a.	speed	presence increase	direct relationship	direct relationship	direct relationship
ь.	pos"	nothing	based on moles	based on enthalpy of rxn	inverse relationship

# **Reaction Rates And Equilibrium Practice Problems Answers**

**N Colangelo** 

# **Reaction Rates And Equilibrium Practice Problems Answers:**

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **Chemistry Workbook For Dummies with Online Practice** Chris Hren, Peter J. Mikulecky, 2017-03-21 Take the confusion out of chemistry with hundreds of practice problems Chemistry Workbook For Dummies is your ultimate companion for introductory chemistry at the high school or college level Packed with hundreds of practice problems this workbook gives you the practice you need to internalize the essential concepts that form the foundations of chemistry From matter and molecules to moles and measurements these problems cover the full spectrum of topics you ll see in class and each section includes key concept review and full explanations for every problem to quickly get you on the right track This new third edition includes access to an online test bank where you ll find bonus chapter guizzes to help you test your understanding and pinpoint areas in need of review Whether you re preparing for an exam or seeking a start to finish study aid this workbook is your ticket to acing basic chemistry Chemistry problems can look intimidating it s a whole new language with different rules new symbols and complex concepts The good news is that practice makes perfect and this book provides plenty of it with easy to understand coaching every step of the way Delve deep into the parts of the periodic table Get comfortable with units scientific notation and chemical equations Work with states phases energy and charges Master nomenclature acids bases titrations redox reactions and more Understanding introductory chemistry is critical for your success in all science classes to follow keeping up with the material now makes life much easier down the education road Chemistry Workbook For Dummies gives you the practice you need to succeed Class 8-12 Chemistry Questions and Answers PDF Arshad Igbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes

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