

Name: _____

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Student Exploration: Reaction Energy

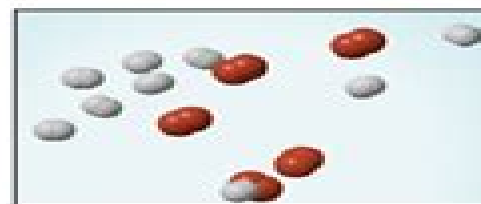
Vocabulary: calorimeter, chemical bond, endothermic, enthalpy, exothermic, Hess's law

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Two magnets are stuck together. What might you have to do to get them to separate?
Pull them apart
2. Suppose you held two magnets a short distance apart, then let go. What would happen?
They would connect
3. Think about the magnets in terms of energy. In which case do you increase the potential energy of the magnets? In which case do you increase the kinetic energy of the magnets?

Gizmo Warm-up

Just like magnets, atoms of different elements are attracted together to form **chemical bonds**. Breaking these bonds requires energy. When a new bond forms, energy is released and temperatures rise. In the *Reaction Energy Gizmo™*, you will explore how the energy of chemical bonding relates to temperature changes that occur during chemical reactions.



To begin, check that **Reaction 1** and **Forward** are selected. In this reaction, hydrogen (H_2) and oxygen (O_2) react to form water (H_2O). The reaction takes place inside a device called a **calorimeter**. Inside the calorimeter, a small chamber holds the reactants. The rest of the calorimeter is filled with water.

1. Click **Play** (▶). What happens? When I clicked play, all of the separate H_2 and O_2 molecules formed to create H_2O .
2. How does the temperature change? The temperature is increased while the molecules are reacting to form water, and the increase in temperature slows once the molecules are all bonded to create the H_2O molecules. The temperature raises from 21.0 Degrees Celsius to 26.0 Degrees Celsius while the molecules were bonding.

Reaction Energy Study Guide Answers

**Thirumalaisamy P.
Velavancorresponding**



Reaction Energy Study Guide Answers:

This book delves into Reaction Energy Study Guide Answers. Reaction Energy Study Guide Answers is an essential topic that must be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Reaction Energy Study Guide Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Reaction Energy Study Guide Answers
- Chapter 2: Essential Elements of Reaction Energy Study Guide Answers
- Chapter 3: Reaction Energy Study Guide Answers in Everyday Life
- Chapter 4: Reaction Energy Study Guide Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Reaction Energy Study Guide Answers. The first chapter will explore what Reaction Energy Study Guide Answers is, why Reaction Energy Study Guide Answers is vital, and how to effectively learn about Reaction Energy Study Guide Answers.

3. In chapter 2, this book will delve into the foundational concepts of Reaction Energy Study Guide Answers. The second chapter will elucidate the essential principles that must be understood to grasp Reaction Energy Study Guide Answers in its entirety.

4. In chapter 3, this book will examine the practical applications of Reaction Energy Study Guide Answers in daily life. The third chapter will showcase real-world examples of how Reaction Energy Study Guide Answers can be effectively utilized in everyday scenarios.

5. In chapter 4, this book will scrutinize the relevance of Reaction Energy Study Guide Answers in specific contexts. The fourth chapter will explore how Reaction Energy Study Guide Answers is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Reaction Energy Study Guide Answers. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Reaction Energy Study Guide Answers.

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