

Name: \_\_\_\_\_ Date: \_\_\_\_\_

## AP Chemistry-Chapter 6 MC Questions

### Multiple Choice

Identify the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. Which periodic group or series of elements is **not** correctly matched with its common family name?
- |                          |                                   |
|--------------------------|-----------------------------------|
| a. alkaline earth metals | IIA                               |
| b. alkali metals         | IIIA                              |
| c. lanthanides           | $_{58}\text{Ce} - _{71}\text{Lu}$ |
| d. halogens              | VIIA                              |
| e. noble gases           | VIIIA                             |
- \_\_\_\_\_ 2. Of the following, which element does **not** match its designation?
- |                     |                         |
|---------------------|-------------------------|
| a. $_{38}\text{Sr}$ | representative metal    |
| b. $_{49}\text{In}$ | representative nonmetal |
| c. $_{14}\text{Si}$ | metalloid               |
| d. $_{74}\text{W}$  | d-transition metal      |
| e. $_{90}\text{Th}$ | f-transition metal      |
- \_\_\_\_\_ 3. Of the following, which periodic group(s) do not match their designation?
- |                          |                        |
|--------------------------|------------------------|
| a. d-transition metals   | IIIB $\rightarrow$ IIB |
| b. representative metals | IA, IIA                |
| c. noble gases           | VIIIB                  |
| d. alkaline earth metals | IIA                    |
| e. halogens              | VIIA                   |
- \_\_\_\_\_ 4. \_\_\_\_\_ is a noble gas.
- |                     |
|---------------------|
| a. $_{22}\text{Ti}$ |
| b. $_{42}\text{Mo}$ |
| c. $_{81}\text{Tl}$ |
| d. $_{36}\text{Kr}$ |
| e. $_{92}\text{U}$  |
- \_\_\_\_\_ 5. \_\_\_\_\_ is an actinide.
- |                     |
|---------------------|
| a. $_{22}\text{Ti}$ |
| b. $_{42}\text{Mo}$ |
| c. $_{81}\text{Tl}$ |
| d. $_{36}\text{Kr}$ |
| e. $_{92}\text{U}$  |
- \_\_\_\_\_ 6. Which one of the following is an inner transition (f-transition) element?
- |       |
|-------|
| a. Rb |
| b. Ho |
| c. Co |
| d. Ru |
| e. Bi |
- \_\_\_\_\_ 7. What would be the outer electron configuration of group IIIB (Sc, Y, La . . .)?
- |                   |
|-------------------|
| a. $ns^2nd^2np^0$ |
| b. $nd^6ns^2$     |
| c. $ns^2(n-1)d^0$ |
| d. $ns^1np^1$     |

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**Huangqi Zhang**



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