National Co.	Theter:	
	L. Carlotte	

9-3 Review and Reinforcement

Classifying chemical Reactions

On the line as the ligh, wide the letter of the type of chamical reaction represented by each numerion below.

 $1. A + B \rightarrow AB$

a, decomposition reaction.

 $_{---}$ 2. A + BX \rightarrow AX + B

4. All + A + B

b. direct combination reaction

- $3. AX + BY \Rightarrow AY + BX$
- double-replacement reaction
 d. single-replacement reaction

Describe in wards each of the following types of reactions.

- 5. direct combination reaction
- 6. decomposition reaction
- T. stagle-replacement reaction.
- N. double-replacement reaction

Decide whether each of the following equations represents a direct combination (dc), decomposition (d), single-replacement (xr), or double-replacement (dr) reaction. Write year univers on the line.

- $\underline{\hspace{1cm}}$ 0. $CO_2 \rightarrow C + O_2$
- _____ ID. NaCl + AgNO₂ → NaNO₂ + AgCl
- $11.5_0 + 401 \rightarrow 8800_0$
- _____12. HaCl₂ + 2NaOH ⇒ 2NaCl + Ra(OH)₂
- _____ 13. Zn + CuSO₁ → ZnSO₁ + Cu
- _____ 14, CH₀ + C + 2H₁
- _____ 15. PbrNO₁u + Mg → Pb + MgtNO₁b

Chairpisters.

Observing Chemical Reactions Reviewand Reinforce

David Baud

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