#### Atomic Number

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

#### Atomic Mass

- \_\_\_\_\_\_ across a period
- down a family

#### Atomic Radius

\_\_\_\_\_ the distance between \_\_\_\_\_ of two like \_\_\_\_\_\_.

- \_\_\_\_\_ across a period (due to increased attraction of more positive nuclei to electron clouds)
- \_\_\_\_\_\_ down a family (due to addition of an energy level)

#### Electron Affinity

amount of \_\_\_\_\_ gains an

- \_\_\_\_\_ across a period
- \_\_\_\_\_ down a family

#### Ionization Energy

amount of \_\_\_\_\_ required to \_\_\_\_\_ an \_\_\_\_ from a neutral \_\_\_\_

- across a period
- \_\_\_\_\_ down a family

#### Electron Distribution - Forming Ions

- Atoms become ions by \_\_\_\_\_ or \_\_\_\_\_ electrons to become.
- An atom is stable when it has a \_\_\_\_\_ outer energy level (\_\_\_\_ valence electrons).
- Atoms that lose (give away) electrons become \_\_\_\_\_ charged ions.
   Atoms that gain (steal) electrons become \_\_\_\_\_ charged ions.

#### The Chemistry Quiz

CR1. CR2. 1. 2. 3. 4. 5.

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John D. Butler

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