

Bohr's Energy Levels

- Electrons in _____
- _____ energy levels: _____ to _____
- _____ energy levels: _____ from _____
- Ground State: _____ in _____ possible

Excited Atom

- Atom has _____
- _____ state is _____
- _____ soon _____ some amount of _____
- _____ seen as _____

Wave Description of Light

Wavelength (_____): _____ between _____ on _____

Frequency (_____): the _____ of _____ waves _____ passing a given _____ in a given _____

$c = \frac{\lambda \nu}{1}$
 $c = \text{_____} : \text{speed of _____}$

Sample problem #1:

What is the frequency of light if the wavelength is $6.0 \times 10^{-7} \text{ m}$?

Note Taking Guide Episode 303 Name Bohrs Energy Levels

SJ Ball



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