

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_6H_{12}O_6$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical $C_6H_{12}O_6$ (molecular)

Formula represents the proportions of _____ of
positive and negative ions in one formula
unit.

Atomic Mass

The proportions that atoms combine through
ions together.

Masses

^{12}C mass 12.011 atomic mass

^{16}O mass 15.999 atomic mass

Subscripts

^{12}C mass atomic mass

^{16}O mass 15.999 atomic mass

^{12}C mass

Note Taking Guide Episode 601 Chemical Formula Answers

Richard Bailey



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