

Chemical Formula

Talking about proportions molecules and relative of atoms indicated by a subscript

Molecular $C_6H_{12}O_6$ (molecules)

Formula represents the number of atoms of each
in a single molecule

Atomic C (atoms)

Formula represents the proportions of
positive and negative ions that combine
ratio

Atomic Mass

The proportions that atoms combine through
ratio

Mass

m mass electrons

m mass proportions through
ratio

Subscript

m mass electrons

m mass proportions through
ratio

m mass

Note Taking Guide Episode 601 Chemical Formula Key

Scott C. Dulebohn



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