

Chemical Formulae

Table 1.4a: number of atoms of each element in a compound

Substance: 1 molecule containing

Formula represents the number of atoms of each element in a molecule

Example 1: 1 molecule containing

Formula represents the number of atoms of protons and neutrons in each molecule

Atomic Number

The number of protons in an atom is equal to the number of neutrons in an atom

Isotopes

- 1. 1 atom with 1 proton and 1 neutron
- 2. 1 atom with 1 proton and 2 neutrons
- 3. 1 atom with 1 proton and 3 neutrons

Relative Atomic Mass

- 1. 1 atom with 1 proton and 1 neutron
- 2. 1 atom with 1 proton and 2 neutrons
- 3. 1 atom with 1 proton and 3 neutrons

Note Taking Guide Episode 601 Answers

Clemens Wendtner



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