# Second Terminal Examination (2022-23)

# Subject- CHEMISTRY

Tim		198	100	THE PERSON NAMED IN
Section 2.	ASSESSED FOR			ALC: UNKNOWN

Clays- 11th Science

Murkst - 70

instructions:-

Question paper consist of 31 questions avaided into four sections namely A. B. C. D.

- 1) Section-A. Q. No. I contains ten multiple choice types of questions carrying one mark each. Q.No.2 contains eight very short answer types of questions carrying one mark each.
- 2) Section-B . Q.No.3 to Q. No.14 are Twelve questions carrying two mark each out of which any eight to be attempted.
- 3) Section-C: Q. No.15 to Q.No.25 are Twelve questions carrying three mark each out of which any eight are to be attempted.
- 4) Section-D: Q No.27 to Q.No. 31 are Five questions carrying four mark each out of which any three to be attempted.

# Section- A

# ().1) Select and write the correct answer:

**CA 400** 

- Which of the following compounds can NOT demonstrate the low of proportions?
  - a) NO.NO:
- b) CO,CO;
- c) H2O, H2O;
- d) Nass, Nat-
- Which of the following temperature will read the same value on Celsius and Fabrenheit Sewies
  - 43 -40

- b) = 40.
- c1-80

- d1 20
- SI unit of the quantity electric current is ...... di Newton c) Candela
  - a) Volts

- b) Ampere

- The no. of molecules in 22.4 cm3 of Nitrogen Gas at STP is IV.
  - a) 6.022 10<sup>20</sup> b) 6.022 10<sup>23</sup> c) 22.4 + 10<sup>20</sup>
- d) 22.4 1022
- The homogeneous series of alcohol has general molecular formula? 1300
- a) Cn Han+OH b) Cn Han OH
- c) Cn Han+ OH

- Find the molar mass of CH<sub>3</sub> CHO. 88.
  - 0144

b) 40

C) 80

d) 30

- Find the molar mass of CH<sub>3</sub> NO<sub>2</sub> SHIRL.
  - a) 6.1

b) 77

01.98

d) 51

- Atomic number of an elements is equal to WHILE.
  - 11) P c-

b) p=N

CIP - A

d) p = n = e

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