



Samsung Ql 6000 Printers Accessory Owners Manual

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temperature change plugging in the values given in the problem you get $q_{\text{water}} = 418 \text{ J/g} \times 110 \text{ g} \times 26.6 \text{ }^{\circ}\text{C}$ $q_{\text{water}} = 550 \text{ J}$ $\Delta h = q_{\text{water}} = 550 \text{ J}$

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