

Use the partial periodic table to answer questions 4 -7.

⁶
C
Carbon
12.0107

— Atomic number
 — Symbol
 — Name
 — Average Atomic Mass

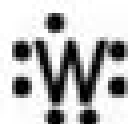
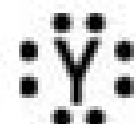
Partial Periodic Table of the Elements

1A	1	H Hydrogen 1.00794
2A	2	He Helium 4.0026
3A	3	Li Lithium 6.941
4A	4	Be Beryllium 9.0122
5A	5	B Boron 10.811
6A	6	C Carbon 12.0107
7A	7	N Nitrogen 14.0064
8A	8	O Oxygen 15.9994
9A	9	F Fluorine 18.9984
10A	10	Ne Neon 20.1797
11A	11	Na Sodium 22.989769
12A	12	Mg Magnesium 24.3040
13A	13	Al Aluminum 26.981538
14A	14	Si Silicon 28.0855
15A	15	P Phosphorus 30.97376
16A	16	S Sulfur 32.06
17A	17	Cl Chlorine 35.453
18A	18	Ar Argon 39.948

1A	2A	3A	4A	5A	6A	7A	8A
1	2	3	4	5	6	7	8
H	He	Li	Be	B	C	N	O
Hydrogen	Helium	Lithium	Beryllium	Boron	Carbon	Nitrogen	Oxygen
1.00794	4.0026	6.941	9.0122	10.811	12.0107	14.0064	15.9994
2	3	4	5	6	7	8	9
Li	Be	B	C	N	O	F	Ne
Lithium	Beryllium	Boron	Carbon	Nitrogen	Oxygen	Fluorine	Neon
6.941	9.0122	10.811	12.0107	14.0064	15.9994	18.9984	20.1797
3	4	5	6	7	8	9	10
Na	Mg	Al	Si	P	S	Cl	Ar
Sodium	Magnesium	Aluminum	Silicon	Phosphorus	Sulfur	Chlorine	Argon
22.989769	24.3040	26.981538	28.0855	30.97376	32.06	35.453	39.948
4	5	6	7	8	9	10	11
K	Ca	Sc	Ti	V	Cr	Mn	Fe
Potassium	Calcium	Scandium	Titanium	Vanadium	Chromium	Manganese	Iron
39.0983	40.078	44.955912	47.867	50.9415	51.9961	54.938045	55.845

Question 5

Suppose scientists discovered four new elements (W, X, Y, Z) while studying rock and soil samples brought back from a Mars mission. Which Lewis dot structure represents an element that should be placed in column VIIA (17) of the periodic table?

- A.  B.  C.  D. 

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G Orfield



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