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 $2\text{ NaCl} + \text{F}_2 \rightarrow 2\text{ NaF} + \text{Cl}_2$
 $2\text{ H}_2\text{O} + 2\text{ H}_2 \rightarrow 2\text{ H}_2\text{O} + 5\text{ H}_2$
 $\text{Pb(OH)}_2 + 2\text{ HCl} \rightarrow 2\text{ H}_2\text{O} + \text{PbCl}_2$
 $2\text{ AlBr}_3 + 3\text{ K}_2\text{SO}_4 \rightarrow 4\text{ KBr} + \text{Al}_2\text{SO}_4 + 3\text{ K}_2\text{SO}_4$
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 $1\text{ O}_2 + 2\text{ NH}_3 \rightarrow \text{HNO}_3 + \text{H}_2\text{O} + 2\text{ O}_2 + 3\text{ H}_2$
 $2\text{ O}_2 + 2\text{ H}_2\text{O} + 2\text{ RbH} + 2\text{ O} + \text{RbOH} + 2\text{ H}_2 + 5\text{ Cl}_2 + 2\text{ KBr} + \text{KCl} + \text{Br}_2 + 6\text{ N}_2 + 2\text{ H}_2 + \text{NH}_3 + 7\text{ Ndf} + 3\text{ Ca} + \text{Nd} + \text{Caf} + 2\text{ H}_2 + 8\text{ CH}_4 + \text{H}_2 + \text{O} + \text{CO} + \text{H}_2 + 9\text{ C} + 2\text{ H}_2 + 5\text{ OH}$

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 $2\text{ FeBr}_3 + 3\text{ H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3\text{ H}_2\text{O} + 2\text{ FeSO}_4$
 $3\text{ C}_4\text{H}_{10} + 13\text{ O}_2 \rightarrow 12\text{ CO}_2 + 10\text{ H}_2\text{O}$

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