

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

- 1. Represent the chemical equation
- 2. Explain how to interpret the equation
- 3. List the steps for writing a chemical equation

Vocabulary

- 1. chemical equation
- 2. chemical equation
- 3. reactant
- 4. coefficient
- 5. balanced equation

Part A Completion

Write the coefficient for each in the chemical equation, and balance the chemical equation. Write the name of each substance in the equation.

A chemical equation is always represented by a symbol _____. The chemical formula for a chemical compound is _____. The law of conservation of mass states that the _____. The number of atoms of each element must be the same on both sides of the equation. A chemical equation is balanced when the number of atoms of each element is the same on both sides of the equation. The chemical equation is balanced when the number of atoms of each element is the same on both sides of the equation.

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Section Review 11 1 Describing Chemical Reaction

Answer Key

Sabine Zange



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