# **How Substances Dissolve**

#### KEY IDEAS

As you read this section, keep these questions in mind:

- · Why is water called the universal solvent?
- · How do substances dissolve?

#### Why Do Substances Dissolve in Certain Solvents?

Water is often called a universal solvent because many substances dissolve in it. However, no one substance can dissolve every solute. A general rule in chemistry is that "like dissolves like." This rule means that a solvent will dissolve substances that have similar molecular structures.

A solvent must be able to attract solute particles and pull them away from one another. Thus, solvent particles must attract solute particles more strongly than the solute particles attract one another.

#### POLAR MOLECULES

Recall that atoms bonded together do not all share electrons equally. For example, a water molecule is made up of two hydrogen atoms bonded to one oxygen atom. The oxygen atom attracts electrons more strongly than the hydrogen atoms do. Thus, the electrons spend more time near the oxygen atom. This gives each hydrogen atom a partial positive charge  $(\delta^*)$  and the oxygen atom a partial negative charge  $(\delta^*)$ .

Molecules that have partially charged positive and negative areas are **polar**. Because water molecules are polar, water can dissolve many other polar substances. Water can also dissolve ionic compounds such as table salt.

Water is a polar molecule because the ongen atom strongly attracts electrons. Because the electrons spend more time near the oxygen atom, the oxygen atom has a partial negative charge.



#### READING TOOLBOX

Summarize Read each page of the section silently to yourself. With a partner, take turns summarizing each page.

# READING CHECK

1. Identify Why is water called a universal solvent?

because

many substances dissolve in it

# W READING CHECK

Explain Why does a water molecule have partial charges?

The oxygen atom attracts electrons more strongly than the hydrogen atoms do. Thus, the electrons spend more time near the oxygen atom

# **Section How Substances Dissolve Answer Key**

**H Kauffman** 

**Section How Substances Dissolve Answer Key:** 

Whispering the Techniques of Language: An Psychological Journey through **Section How Substances Dissolve Answer Kev** 

In a digitally-driven earth wherever screens reign great and quick interaction drowns out the subtleties of language, the profound secrets and psychological nuances hidden within words frequently move unheard. However, situated within the pages of **Section How Substances Dissolve Answer Key** a captivating literary prize pulsating with natural thoughts, lies an extraordinary journey waiting to be undertaken. Published by a talented wordsmith, this marvelous opus encourages readers on an introspective journey, softly unraveling the veiled truths and profound affect resonating within the very cloth of each and every word. Within the mental depths with this emotional evaluation, we shall embark upon a honest exploration of the book is core themes, dissect its fascinating publishing design, and yield to the effective resonance it evokes deep within the recesses of readers hearts.

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