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NJames IDva be Section 2-3 Carbon Compounds (pages 44-48) Key Concept What are the functions of each group of organic compounds? The Chemistry of Carbon (page 44) How many valence electrons does each carbon atom have? Each carbon atom has four electrons. 2. What gives carbon the ability to form chains that are almost unlimited in length? A carbon atom can bond to other carbon atoms. Macromolecules (page 45) Many of the molecules in living cells are so large that they are known as rmacromolecules. What is the process called by which macromolecules are formed? Polymerization When monomers join together, what do they form? Polymers 6. What are four groups of organic compounds found in living things? Carbohydrates b. Lipids Nucleic acids d. Proteins Carbohydrates (pages 45-46) 7. What atoms make up carbohydrates? Carbon, hydrogen, and oxygen atoms make up carbolrydrates. Circle the letter of each sentence that is true about carbohydrates. (a.) Starches and sugars are examples of carbohydrates. (b) Living things use them as their main source of energy. The monomers in sugar polymers are starch molecules.

(d.) Plants and some animals use them for strength and rigidity.

Single sugar molecules are also called <u>monosaccharides</u>

Circle the letter of each monosaccharide.

(a.) galactose

c.) glucose

b. glycogen

(d.) fructose

Section Carbon Compounds Answers

KJ Lindholm-Leary

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