

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 & \times & .692 = 43.596 \rightarrow 44 \\ 65 & \times & .308 = 20.02 \rightarrow 20 \\ & + & \\ & \hline & & 63.62 \rightarrow 64 \end{array}$$

Section 7 1 Review Answers Chemistry

Marco Cascella



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CliffsNotes HESI A2 Science Cram Plan Michael Reid,2021-04-13 A study guide for the HESI A2 science nursing school test that calendarizes a study plan for test takers depending on how much time they have left before taking the test

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone,Theodore O. Dolter,2012-01-03 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter A new Math Check allows quick access to the needed basic skill The first chapter now includes brief introductions to several fundamental chemical concepts and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter Chemistry 'O' Level Rex M. Heyworth,2007

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additional interactive questions **Chemistry insights 'O' level** Rex M. Heyworth,2007 Modern Chemistry Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2001 **Linne & Ringsrud's Clinical Laboratory Science E-Book** Mary Louise Turgeon,2018-12-22 Thoroughly updated and easy to follow Linne Ringsrud s Clinical Laboratory Science Concepts Procedures and Clinical Applications 8th Edition offers a fundamental overview of the laboratory skills and techniques you ll need for success in the clinical laboratory Author Mary Louise Turgeon s simple and straightforward writing clarifies complex concepts and her unique discipline by discipline approach helps you build knowledge and learn to confidently perform routine clinical laboratory tests with accurate effective results Topics like safety measurement techniques and quality assessment are woven throughout the various skills The new eighth edition also features updated content including expanded information on viruses and automation It s the must have foundation for anyone wanting to pursue a profession in the clinical lab Broad content scope provides an ideal introduction to clinical laboratory science at a variety of levels including CLS MT CLT MLT and Medical Assisting Case studies include critical thinking and multiple choice questions to challenge readers to apply the content to real life scenarios Expert insight from respected educator Mary Lou Turgeon reflects the full spectrum of clinical lab science Detailed procedures guides readers through the exact steps performed in the lab Vivid full color illustrations familiarize readers with what they ll see under the microscope Review questions at the end of each chapter help readers assess your understanding and identify areas requiring additional study Evolve companion website provides convenient online access to all of the procedures in the text and houses animations flashcards and additional review questions not found in the printed text Procedure worksheets can be used in the lab and for assignment as homework Streamlined approach makes must know concepts and practices more accessible Convenient glossary simplifies the process of looking up definitions without having to search through each chapter NEW Updated content throughout keeps pace with constant changes in clinical lab science NEW Consistent review question format ensures consistency and enables readers to study more efficiently NEW More discussion of automation familiarizes readers with the latest automation technologies and processes increasingly used in the clinical lab to increase productivity and elevate experimental data quality NEW Additional information on viruses keeps readers up to date on this critical area of clinical lab science **Sif Chemistry Nl Tb** Rex M. Heyworth,2007 **CliffsNotes AP Chemistry** Bobrow Test Preparation Services,2009-02-09 The book itself contains chapter length subject reviews on every subject tested on the AP Chemistry exam as well as both sample multiple choice and free response questions at each chapter s end Two full length practice tests with detailed answer explanations are included in the book **The Independent** ,1919 **Biology for the Informed Citizen** Donna M. Bozzone,Douglas S. Green,2014 Biology for the Informed Citizen helps student connect the concepts of biology to the consequences of biology This text aims to teach the concepts of biology evolution and the process of science so students can apply this knowledge in their everyday lives as informed consumers and users of scientific informationThis

version of the text does not feature Physiology For more information about Biology for the Informed Citizen with Physiology please search for ISBN 9780195381993 Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2016-08-02 This text is an unbound three hole punched version Used by over 750 000 students Foundations of College Chemistry Binder Ready Version 15th Edition is praised for its accuracy clear no nonsense approach and direct writing style Foundations direct and straightforward explanations focus on problem solving making it the most dependable text on the market Its comprehensive scope proven track record outstanding in text examples and problem sets were all designed to provide instructors with a solid text while not overwhelming students in a difficult course Foundations fits into the prep intro chemistry courses which often include a wide mix of students from science majors not yet ready for general chemistry allied health students in their 1st semester of a GOB sequence science education students for elementary school teachers to the occasional liberal arts student fulfilling a science requirement Foundations was specifically designed to meet this wide array of needs Green Organic Chemistry and its Interdisciplinary Applications Vera M. Kolb, 2017-04-21 Green Organic Chemistry and Its Interdisciplinary Applications covers key developments in green chemistry and demonstrates to students that the developments were most often the result of innovative thinking Using a set of selected experiments all of which have been performed in the laboratory with undergraduate students it demonstrates how to optimize and develop green experiments The book dedicates each chapter to individual applications such as Engineering The chemical industry The pharmaceutical industry Analytical chemistry Environmental chemistry Each chapter also poses questions at the end with the answers included By focusing on both the interdisciplinary applications of green chemistry and the innovative thinking that has produced new developments in the field this book manages to present two key messages in a manner where they reinforce each other It provides a single and concise reference for chemists instructors and students for learning about green organic chemistry and its great and ever expanding number of applications **McGraw-Hill's SAT Subject Test Chemistry, 3rd Edition** Thomas Evangelist, 2011-02-03 Expert guidance on the Chemistry exam Many colleges and universities require you to take one or more SAT II Subject Tests to demonstrate your mastery of specific high school subjects McGraw Hill s SAT Subject Test Chemistry is written by experts in the field and gives you the guidance you need perform at your best This book includes 4 full length sample tests updated for the latest test format 40 top tips to remember on test day Glossary of tested chemistry terms and formulas Tips and strategies from one of the most popular teachers at the renowned Brooklyn Technical High School Diagnostic test to pinpoint strengths and weaknesses Step by step review of all topics covered on the exam In depth coverage of the lab experiment questions that are a major test feature Charts tables and illustrations to simplify and reinforce learning Practice tests just like the real SAT Subject Test in Chemistry Test taking tips and strategies *Biophysical Chemistry* Charles R. Cantor, Paul R. Schimmel, 1980-04-15 Three part series remains the definitive text on the physical properties of biological macromolecules and the physical techniques used to study them It is

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