

## Chemistry of Life

# Study Guide B

## Answer Key

### SECTION 1. ATOMS, IONS, AND MOLECULES

1. An element is a certain type of atom.
2. Sketch should resemble one of the illustrations in Figure 1.2 in Section 1. Nucleus should be in the center with protons and neutrons. Electrons should be labeled in a ring surrounding the nucleus.
3. A compound is composed of atoms of different elements.
4. an atom that has gained or lost one or more electrons
5. the number of protons no longer equals the number of electrons
6. Sketch should resemble the first illustration in Figure 1.3, with one atom losing an electron (becoming a positive ion) and the other accepting that electron (becoming a negative ion).
7. A bond that forms between two atoms that share a pair of electrons.
8. the number of electron pairs that it needs to share to fill its outer energy level
9. molecule
10. compound
11. element
12. ion
13. Ionic bonds form between oppositely-charged ions; covalent bonds form when atoms share a pair of electrons.

### SECTION 2. PROPERTIES OF WATER

1. a molecule with a slightly positively charged region and a slightly negatively charged region
2. The oxygen nucleus pulls the electrons in the molecule more strongly than the hydrogen atoms. The hydrogen atoms have a slight positive charge and the oxygen atom has a slight negative charge.
3. an attraction between a slightly positive hydrogen and a slightly negative atom
4. between the slightly positive hydrogen atom of one water molecule and the slightly negative oxygen atom of another water molecule
5. A large amount of energy is needed to produce an increase in water temperature; water resists changes in temperature.
6. Water molecules "stick" to each other.
7. Water molecules "stick" to other substances.
8. A solvent is present in greater concentration and dissolves the solute.
9. ions and polar molecules
10. nonpolar molecules, such as fats and oils
11. **Effect on  $H^+$  concentration:** acids increase  $H^+$  concentration; bases decrease  $H^+$  concentration  
**Effect on pH:** acids lower pH; bases raise pH
12. Sketch should resemble Visual Vocab in Section 2 of the text. Solution: homogeneous mixture of substances; solvent: substance present in greater amount in which other substances dissolve; solute: substance that dissolves in a solvent.

### SECTION 3. CARBON-BASED MOLECULES

1. Carbon atoms are the basis of the molecules that make up most living things.
2. Carbon atoms can form covalent bonds with up to four other atoms, including other carbon atoms.
3. Sketches should resemble Figure 3.1 in Section 3.
4. broken down as a source of chemical energy; part of cell structure
5. sugars, starches, cellulose
6. broken down as a source of chemical energy; part of cell structure
7. fats, oils, phospholipids
8. many functions, including movement, transport, chemical catalysts
9. enzymes, hemoglobin
10. store genetic information, build proteins

# Section 1 Study Guide B Answers

**Camilla Rothe**



## Section 1 Study Guide B Answers:

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### **Table of Contents Section 1 Study Guide B Answers**

1. Understanding the eBook Section 1 Study Guide B Answers
  - The Rise of Digital Reading Section 1 Study Guide B Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Section 1 Study Guide B Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Section 1 Study Guide B Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Section 1 Study Guide B Answers
  - Personalized Recommendations
  - Section 1 Study Guide B Answers User Reviews and Ratings
  - Section 1 Study Guide B Answers and Bestseller Lists



5. Accessing Section 1 Study Guide B Answers Free and Paid eBooks
  - Section 1 Study Guide B Answers Public Domain eBooks
  - Section 1 Study Guide B Answers eBook Subscription Services
  - Section 1 Study Guide B Answers Budget-Friendly Options
6. Navigating Section 1 Study Guide B Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Section 1 Study Guide B Answers Compatibility with Devices
  - Section 1 Study Guide B Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Section 1 Study Guide B Answers
  - Highlighting and Note-Taking Section 1 Study Guide B Answers
  - Interactive Elements Section 1 Study Guide B Answers
8. Staying Engaged with Section 1 Study Guide B Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Section 1 Study Guide B Answers
9. Balancing eBooks and Physical Books Section 1 Study Guide B Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Section 1 Study Guide B Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Section 1 Study Guide B Answers
  - Setting Reading Goals Section 1 Study Guide B Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section 1 Study Guide B Answers
  - Fact-Checking eBook Content of Section 1 Study Guide B Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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