

## 19

## ACIDS, BASES, AND SALTS

**SECTION 19.1 ACID-BASE THEORIES (Pages 587-593)**

This section compares and contrasts acids and bases as defined by the theories of Arrhenius, Brønsted-Lowry, and Lewis. It also introduces conjugate acid-base pairs in acid-base reactions.

**► Properties of Acids and Bases (pages 587-588)**

1. Circle the letters of all the terms that complete the sentence correctly.

The properties of acids include \_\_\_\_\_.

- a. reacting with metals to produce oxygen
- b. giving foods a sour taste
- c. forming solutions that conduct electricity
- d. causing indicators to change color

2. Bases are compounds that react with acids to form \_\_\_\_\_ and a(n) \_\_\_\_\_.

3. Circle the letters of all the terms that complete the sentence correctly.

The properties of bases include \_\_\_\_\_.

- a. tasting bitter
- b. feeling slippery
- c. changing the color of an indicator
- d. always acting as a strong electrolyte

4. What are the three main definitions of acids and bases? \_\_\_\_\_

5. Complete the table about the three definitions of acids and bases. (pg 592)

| Acid-Base Definitions |                         |                         |
|-----------------------|-------------------------|-------------------------|
| Type                  | Acid                    | Base                    |
| Arrhenius-Lowry       |                         | H <sup>+</sup> acceptor |
|                       | electron-pair acceptor  |                         |
|                       | H <sup>+</sup> producer |                         |

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**E Durkheim**



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