



## Neutralization Reactions Worksheet

- 1. Write the balanced chemical equations for the neutralization reactions between:
  - a) HI and NaOH

$$HI + NaOH \rightarrow H_2O + NaI$$

b) H<sub>2</sub>CO<sub>3</sub> and Sr(OH)<sub>2</sub>

$$H_2CO_3 + Sr(OH)_2 \rightarrow 2 H_2O + SrCO_3$$

c) Ca(OH)<sub>2</sub> and H<sub>3</sub>PO<sub>4</sub>

$$3 \text{ Ca}(OH)_2 + 2 \text{ H}_3PO_4 \rightarrow 6 \text{ H}_2O + \text{Ca}_3(PO_4)_2$$

d) hydrobromic acid and barium hydroxide

e) zine hydroxide and nitrie acid

$$Zm(OH)_2 + 2 HNO_3 \rightarrow 2 H_2O + Zm(NO_3)_2$$

aluminum hydroxide and hydrochloric acid

$$AI(OH)_3 + 3 HCI \rightarrow 3 H_2O + AICI_3$$

2. Complete and balance the following equations representing neutralization reactions:

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**BM King** 

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Reviewing Section 19 4 Neutralization Reactions Worksheet Amswers: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "Section 19 4 Neutralization Reactions Worksheet Amswers," an enthralling opus penned by a highly acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve into the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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