

Section 12.3 Limiting Reagent & Percent Yield

Answer Key

Michelle Galian

Vocabulary

1. e
2. c
3. d
4. a
5. b

1. Answer: 127g Cu
Because...

Find the limiting reagent:

$$54\text{g Al} \times 1 \text{ mole Al}/27\text{g Al} = 2 \text{ mole Al}$$

$$319\text{g CuSO}_4 \times 1 \text{ mole CuSO}_4/159.5\text{g CuSO}_4 = 2 \text{ mole CuSO}_4 \text{ (limiting reagent)}$$

$$2 \text{ mole CuSO}_4 \times 3 \text{ mole Cu}/3 \text{ mole CuSO}_4 \times 63.5\text{g Cu}/1 \text{ mole Cu} = 127\text{g Cu}$$

(Multiply the founded limiting reagent with moles of Copper (3) over moles of CuSO₄ (3) from the equation to cancel out moles of CuSO₄.)

Then, to cancel out 3 moles Cu, multiply it with grams Cu (because that is also what the question asks for) over 1 mole Cu. After canceling and multiplying out what is left, you come to 127g Cu.)

2. $27\text{g Cu} \times 1 \text{ mole Cu}/63.5\text{g Cu} \times 2 \text{ mole Cr}/3 \text{ mole Cu} \times 52\text{g Cr}/1 \text{ mole Cr} = 14.74\text{g Cr}$

("excess of CuSO₄" is irrelevant to the problem you are solving, so you can cross that out. You are left only to work with 27g Cu, and need to find grams Cr. So you divide 27g Cu by its molar mass [from periodic table] and then multiply it by the mole to mole ratio of 2 moles Cr over 3 moles Cu. You put 3 moles Cu on the bottom to cancel it out with 1 mole Cu. Lastly, to cancel out 2 moles Cr, multiply it with the molar mass [52g] over 1 mole Cr and multiply out what is uncrossed.)

Section 12 3 Limiting Reagent And Percent

LP Steffe



Section 12 3 Limiting Reagent And Percent:

Unveiling the Magic of Words: A Review of "**Section 12 3 Limiting Reagent And Percent**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Section 12 3 Limiting Reagent And Percent**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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