



Science Curriculum

Chemical Changes

Read the passage and answer the following questions.



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in the medical field as a disinfectant and antiseptic. The reaction of hydrogen peroxide to form water and oxygen is a common example of a chemical change.

4. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, what is the mass of the products compared to the mass of the reactants?

It is equal.



5. What is the chemical formula for water? Is it a liquid, solid, or gas?

6. What is the chemical formula for oxygen? Is it a liquid, solid, or gas?

7. What is the chemical formula for hydrogen peroxide? Is it a liquid, solid, or gas?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in the medical field as a disinfectant and antiseptic.

8. What is the chemical formula for water? Is it a liquid, solid, or gas?

9. What is the chemical formula for oxygen? Is it a liquid, solid, or gas?

Hydrogen peroxide is a colorless liquid with a strong, sharp odor. It is a powerful oxidizing agent and is used in many industrial and household applications. It is also used in the medical field as a disinfectant and antiseptic.

10. According to the law of conservation of mass, if hydrogen peroxide decomposes into water and oxygen, what is the mass of the products compared to the mass of the reactants?

Section Reinforcement Chemical Changes Answer Key

R Barnett



Section Reinforcement Chemical Changes Answer Key:

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