# Ponic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms Valence electrons Electron transfer from/to each atom
Li Ive Housters Ive to F.
F Tye accepts Ive from Li F

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	loss formed in the product
Li +ci→ Lici  Li +ci→ Lici  Li +ci→ Lici	Li	Ive	donate le-	Lit
	CI	Ive	accepts 1e	C1-1
2) Ca+0=> Ca0  (a) + 1 - 0: -> [a] +2 [0:] 2	2 Ca	Dre	Starch 205	Cata
	0	6ve	accepts	0-
Be F: BeF:	-1 Re	Que	donates de to	Beto
	F	712	each F accepts Ive each	
mg + 5 → [mg]+2[:S:]2	2 mg	Sve	denotes ae to	wet2
	S	Gve	assepts a e	5->
S)K+F⇒KF  KU+FE: → [K]+[E]	K	Ivē	donates	K+I
	=	7ve	accepts	F-/

# **Section 6 Ionic Bonding Worksheet Answer Key**

**JR Anderson** 

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