

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Section 6 1 Review Chemical Bonding Answer Key

Y Pai

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

Section 6 1 Review Chemical Bonding Answer Key:

Enjoying the Track of Term: An Mental Symphony within **Section 6 1 Review Chemical Bonding Answer Key**

In a world taken by screens and the ceaseless chatter of fast conversation, the melodic elegance and psychological symphony developed by the written word usually diminish in to the back ground, eclipsed by the persistent noise and disruptions that permeate our lives. However, nestled within the pages of **Section 6 1 Review Chemical Bonding Answer Key** a marvelous literary treasure filled with organic emotions, lies an immersive symphony waiting to be embraced. Crafted by an elegant musician of language, this captivating masterpiece conducts viewers on a mental trip, skillfully unraveling the hidden tunes and profound affect resonating within each cautiously constructed phrase. Within the depths of this poignant evaluation, we can discover the book is central harmonies, analyze their enthralling publishing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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