

1. In the list below, circle the properties of acids and box the properties of bases.

Sour taste	Feel slippery	Contains hydroxide ions
Turns litmus blue	Reacts with metals to form H_2	Turns litmus red
Bitter taste	Contains hydrogen ions	Corrosive

2. Complete the chart below:

Acid-Base Definitions		
Type	Acid Definition	Base Definition
Arrhenius	Donates an H^+ ion in solution	Donates an OH^- ion in solution
Bronsted-Lowry	Donates an H^+ ion	Accepts an H^+ ion
Lewis	Accepts an electron pair	Donates an electron pair

3. What term(s) describe a substance that can act as an acid or a base? **Amphoteric or amphiprotic**
4. Neutralization reactions always produce a **SALT** and **WATER**
5. If the $[H^+]$ is greater than the $[OH^-]$, the solution is acidic basic neutral
6. If the $[OH^-]$ is greater than the $[H^+]$, the solution is acidic basic neutral
7. If the $[H^+]$ is equal to the $[OH^-]$, the solution is acidic basic neutral
8. If the pH is 7, the solution is acidic basic neutral
9. If the pH is greater than 7, the solution is acidic basic neutral
10. If the pH is less than 7, the solution is acidic basic neutral
11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?



$$(19.2 \text{ g } HClO_3) \times (1 \text{ mol } HClO_3 / 84 \text{ g } HClO_3) \times (1 \text{ mol } LiOH / 1 \text{ mol } HClO_3) = \underline{0.228 \text{ mol } LiOH}$$

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?



$$(132.4 \text{ g } HNO_3) \times (1 \text{ mol } HNO_3 / 63 \text{ g } HNO_3) \times (1 \text{ mol } Ca(OH)_2 / 2 \text{ mol } HNO_3) \times (74 \text{ g } Ca(OH)_2 / 1 \text{ mol } Ca(OH)_2) = \underline{77.76 \text{ g } Ca(OH)_2}$$

Section 20 Acids And Bases Answers

**DM Vasudevan, Sreekumari S, Kannan
Vaidyanathan**



Section 20 Acids And Bases Answers:

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successful problem solving in organic chemistry Existing textbooks provide extensive coverage of the principles but there is far less emphasis on the skills needed to actually solve problems

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Acid-base Interactions K. L. Mittal, This book documents the proceedings of the Second International Symposium on Acid Base Interactions Relevance to Adhesion Science and Technology held in Newark New Jersey October 19 21 1998 Since the first symposium on this topic was held on the occasion of the 75th birthday of Professor Frederick M Fowkes in 1990 it was deemed opportune and necessary to hold the second symposium on this topic This symposium was organized with the following objectives in mind i to consolidate the R Part 2 Characterization of the Acid Base Properties of Materials and Part 3 Applications of Acid Base Interactions The topics covered include Surface free energy acid base theory applied to solid surfaces Good van Oss and Chaudhury theory contact angle measurements and interpretation acid base theory of contact angles acid base strength of solid surfaces acid base interactions at solid surfaces acid base interactions at the molecular level characterization of acid base properties of a host of materials polymers wood glass ceramics silica particles textile fibers rocks by XPS inverse gas chromatography immersion calorimetry contact angle titration and thin layer wicking and relevance of acid base interactions to bioadhesion microbial adhesion polymer adhesion and adhesion in reinforced polymer composites

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