

Matter—Properties and Changes

Section 3.1 Properties of Matter

pages 70–75

Problem-Solving Lab

1. Explain why the flow of a compressed gas must be controlled for practical and safe use.

The flow of compressed gas must be controlled to control the amount and the rate at which gas is released.

2. Predict what would happen if the valve on a full tank of compressed gas were suddenly opened all the way or if the tank were accidentally punctured.

Without the regulator device, the gas would rush out of the tank with a force powerful enough to transform the tank into a dangerous, uncontrolled projectile.

Section 3.1 Assessment

page 75

1. Create a table that describes the three common states of matter in terms of their shape, volume, and compressibility.

	Volume	Shape	Compressibility
Solid	Definite	Definite	Incompressible
Liquid	Definite	Takes shape of container and fills container to the extent of its own volume	Virtually incompressible
Gas	Fills volume of container	Takes shape of container	Compressible

2. Describe the characteristics that identify a sample of matter as being a substance.

The sample of matter must have a uniform and unchanging composition to be a substance.

3. Classify each of the following as a physical or chemical property.

- a. Iron and oxygen form rust.
chemical

- b. Iron is more dense than aluminum.

physical

- c. Magnesium burns brightly when ignited.

chemical

- d. Oil and water do not mix.

physical

- e. Mercury melts at -39°C .

physical

4. **Organize** Create a chart that compares physical and chemical properties. Give two examples for each type of property.

The chart should make clear that physical properties can be observed without changing the composition of the sample, which is not the case for chemical properties. Mass and density are examples of physical properties. Fermentation and rusting are examples of chemical properties.

Section 3.2 Changes in Matter

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Practice Problems

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5. Use the data in the table to answer the following questions.

Aluminum and Liquid Bromine Reaction

	Before Reaction	After Reaction
Aluminum	10.3 g	0.0 g
Liquid bromine	100.0 g	8.5 g
Compound	0.0 g	

How many grams of bromine reacted? How many grams of compound were formed?

amount of bromine that reacted =
 $100.0\text{ g} - 8.5\text{ g} = 91.5\text{ g}$

amount of compound formed =
 $100.0\text{ g} + 10.3\text{ g} - 8.5\text{ g} = 101.8\text{ g}$

Section 2 Matter Answers

Rachel Sandford



Section 2 Matter Answers:

Solutions Manual for Chemistry: Molecules Matter and Change, Fourth Edition Julie Henderleiter, Charles Trapp, Peter Atkins, Loretta Jones, 1999-08-13 This student companion is a supplement to Chemistry Molecules Matter and Change 4th edition with CD ROM It features guided reading strategies collaborative learning sheets and strategies for using CD ROM tools *Complete Homeschool Social Science* Terri Raymond, 2015-03-05 This book is a collection of Home School Brews bestselling social science series It covers grades 1 to 6 Each book may also be purchased separately *Amend Section 2 of the Export Control Act of 1949, Hearings Before a Subcommittee of ..., 89-1 on S.948 ..., May 24 and 25, 1965* United States. Congress. Senate. Committee on Banking and Currency, 1965

Decisions and Orders of the National Labor Relations Board United States. National Labor Relations Board, 2008 *The Coal Trade Bulletin*, 1921 **Union Postal Clerk**, 1910

The Union Postal Clerk George A. Donnelly, 1911 United States Code United States, 1950 **Traffic World and Traffic Bulletin**, 1929 **Federal Register**, 2013-06 **Instantons in Gauge Theories** M. Shifman, 1994 This volume is a compilation of works which taken together give a complete and consistent presentation of instanton calculus in non Abelian gauge theories as it exists now Some of the papers reproduced are instanton classics Among other things they show from a historical perspective how the instanton solution has been found the motivation behind it and how the physical meaning of instantons has been revealed Other papers are devoted to different aspects of instanton formalism including instantons in supersymmetric gauge theories A few unsolved problems associated with instantons are described in great detail The papers are organized into several sections that are linked both logically and historically accompanied by extensive comments

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State Implementation of Federal Standards United States. Congress. Senate. Committee on Governmental Affairs. Subcommittee on Intergovernmental Relations,1982 *Justice of the Peace and Local Government Review* ,1840 Quantum Physics of Matter Alan Durrant,2000-01-01

Quantum Physics of Matter explores the way in which quantum physics determines the properties of materials The quantum physics of solids for example dictates whether they are good insulators conductors semiconductors or even superconductors At a deeper level it explores how the quantum physics of nuclei and elementary particles determines the stability of matter and hence the range of substances that came into existence through the big bang and the evolution of stars **Code of Federal Regulations** ,1996 2018 CFR Annual

Print Title 17 Commodity and Securities Exchanges Parts 1 to 40 Office of The Federal Register,2018-04-01

Congressional Record United States. Congress,1954 The Congressional Record is the official record of the proceedings and debates of the United States Congress It is published daily when Congress is in session The Congressional Record began publication in 1873 Debates for sessions prior to 1873 are recorded in The Debates and Proceedings in the Congress of the United States 1789 1824 the Register of Debates in Congress 1824 1837 and the Congressional Globe 1833 1873

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