

PROBLEMLÖSUNG

Seite 4

Prüfung: 1. September 2024

BEWEIS

Gegeben: 100 g festes Natriumperoxyd wird mit 100 g festem Natriumperoxyd gemischt.

Bestimmen Sie die Masse des Natriumperoxyds, das bei der Reaktion entsteht.

Gegeben: 100 g festes Natriumperoxyd

$2Na_2O_2 \rightarrow 2Na_2O + O_2$

$2 \cdot 78 \text{ g} = 156 \text{ g}$

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Gegeben: 100 g festes Natriumperoxyd

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Gegeben: 100 g festes Natriumperoxyd, 100 g festes Natriumperoxyd, 100 g festes Natriumperoxyd

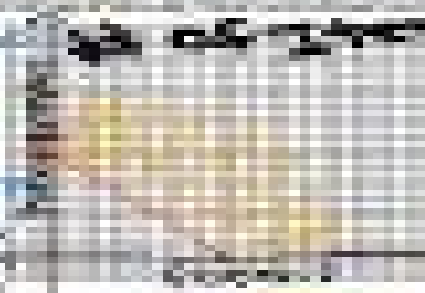
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Gegeben: 100 g festes Natriumperoxyd, 100 g festes Natriumperoxyd, 100 g festes Natriumperoxyd

Secondary Math 1 Module 5

Mathematics Project



Secondary Math 1 Module 5:

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NASA Contractor Report ,1968 *Teaching in the Middle and Secondary Schools* Joseph F. Callahan,Leonard H. Clark,Richard Dean Kellough,1998 This book is designed to provide a practical criterion referenced performance based model for general secondary or middle school methods The text s unique modular format allows students to work through each module at their own pace building confidence as each skill is mastered Using current and practical methods of teaching this latest edition incorporates constructivist based theory social interactive learning strategies and inquiry learning which involve students in problem solving situations

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