# Graphing Periodic Trends Lab

Name Hr\_

PURPOSE: To determine how certain properties are periodic when the elements are arranged in periods or groups.

DISCUSSION: There are many periodic trends that occur as you cross a period or go down a group/family on the Periodic Table. Graphing various elemental properties will help to illustrate these trends.

#### REFERENCE:

Angstrom - A unit of distance equal to one ten-billionth of a meter.

Atomic Radius - The distance from the center of the nucleus to the valence electron energy level.

ionization energy - The energy required to remove an electron from an atom.

#### DATA

Properties of the First 18 Elements

| Element | Atomic<br>Number | (Angstroms) | Ionization Energy |
|---------|------------------|-------------|-------------------|
| н       | 1                | 0.79        | 314               |
| He      | 2                | 0.49        | 567               |
| Li      | 3                | 2.06        | 124               |
| Ве      | 4                | 1.4         | 215               |
| 8       | 5                | 1.17        | 191               |
| C       | 6                | 0.91        | 260               |
| H       | 7                | 0.75        | 335               |
| 0       | 8                | 0.65        | 314               |
| F       | 9                | 0.57        | 402               |
| No      | 10               | 7           | 2                 |
| Ma      | 11               | 2.23        | 119               |
| Mg      | 12               | 1.72        | 176               |
| A.I     | 13               | 1.82        | 138               |
| Si      | 1.4              | 1.46        | 169               |
| P       | 15               | 1.23        | 242               |
| S       | 16               | 1.09        | 239               |
| CI      | 17               | 0.97        | 299               |
| Ar      | 18               | 2           | 2                 |

Using the data above, make a line graph the following data. Draw a dark vertical line to separate the 3 periods. Give each graph a title and label both the x-axis and the y-axis. Connect your points. Predict the plot points for elements with guestions marks.

- Atomic Radius (y) vs. Atomic Number (x) for the first 3 periods (18 elements 3 graphs)
- Ionization energy (y) vs. Atomic Number (x) for the first 3 periods (18 elements 3 graphs)

## **Periodic Trends Graphing Lab Answers**

**Hussin A.Rothana** 

### **Periodic Trends Graphing Lab Answers:**

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