

PERIODIC TABLE TRENDS

Step 1: Complete the squares for each element by adding the atomic number, name, and atomic mass.

Write the atomic number at the top of the square.

Write the atomic name under the symbol.

Write the atomic mass at the bottom of the square.

Step 2: Determine the number of electrons, protons, and neutrons in each statement.

Step 3: Create a Bohr diagram for each element.

Step 4: Draw a Lewis Dot Structure for each element.

Step 5: Show whether the element is a metal, nonmetal, or metalloid by circling the correct response. Also determine if the element is a solid or gas by circling the correct choice.

Step 6: Use the following colors to shade in the square for each element. You should **ONLY** color in the small square in the upper left-hand corner and not the entire card.

Green – Li and Na

Pink – O and S

Blue – Be and Mg

Purple – F and Cl

Orange – B and Al

Red – C and Si

Brown – N and P

Yellow – He, Ne, and Ar

White – H

Step 7: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

Step 8: Put a title on your table.

Step 9: Make a key for each color with the name of the group for that column.

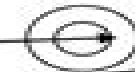
Step 10: Write the column number at the top of each group and the period number on the left side of the table. Be careful, column numbers are not in order from 1-18 and the transition metals, columns 3-12 are not on the table.

Step 11: Draw the red zigzag line that separates the metals from the nonmetals.

Step 12: Answer the questions on the worksheet using the information on your periodic table.

Be

Bohr Diagram



Metal Nonmetal Metalloid

Solid Gas

P = ____

N = ____

E = ____

Be

Section Trends In The Periodic Table Answer Key

Oswaal Editorial Board



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Table of Contents Section Trends In The Periodic Table Answer Key

1. Understanding the eBook Section Trends In The Periodic Table Answer Key
 - The Rise of Digital Reading Section Trends In The Periodic Table Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Section Trends In The Periodic Table Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section Trends In The Periodic Table Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section Trends In The Periodic Table Answer Key
 - Personalized Recommendations
 - Section Trends In The Periodic Table Answer Key User Reviews and Ratings
 - Section Trends In The Periodic Table Answer Key and Bestseller Lists
5. Accessing Section Trends In The Periodic Table Answer Key Free and Paid eBooks
 - Section Trends In The Periodic Table Answer Key Public Domain eBooks
 - Section Trends In The Periodic Table Answer Key eBook Subscription Services
 - Section Trends In The Periodic Table Answer Key Budget-Friendly Options
6. Navigating Section Trends In The Periodic Table Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Section Trends In The Periodic Table Answer Key Compatibility with Devices
- Section Trends In The Periodic Table Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section Trends In The Periodic Table Answer Key
 - Highlighting and Note-Taking Section Trends In The Periodic Table Answer Key
 - Interactive Elements Section Trends In The Periodic Table Answer Key
- 8. Staying Engaged with Section Trends In The Periodic Table Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section Trends In The Periodic Table Answer Key
- 9. Balancing eBooks and Physical Books Section Trends In The Periodic Table Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section Trends In The Periodic Table Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Section Trends In The Periodic Table Answer Key
 - Setting Reading Goals Section Trends In The Periodic Table Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Section Trends In The Periodic Table Answer Key
 - Fact-Checking eBook Content of Section Trends In The Periodic Table Answer Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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