

Use your periodic table to fill in the missing spaces below. Consider all atoms to be stable.

Element	Symbol	Atomic Number	Mass Number	Protons	Neutrons	Electrons
Carbon		6	12			
	O	8			8	
Potassium				19	20	
	Au					
Tin	Sn	50			69	
Zinc			64	30		
Cobalt	Co	27			32	
Boron				5	6	
	Fe					

Use your periodic table to complete the table below. The first one has been done for you.

Element	Atomic Number	Mass Number	Protons	Neutrons	Electrons	Symbol
Iodine	53	127	53	74	54	I^{-1}
		35	17			Cl^{-1}
Hydrogen		1	1		0	
	35			45	36	
	5	11				B^{+3}
Oxygen		16	8		10	
Nitrogen			7	7	10	
Aluminum		27			10	
		56	26		24	
	29			34		Cu^{+1}
		63	29		27	

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D Keegan

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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