

11.1

DESCRIBING CHEMICAL REACTIONS

Section Review

Objectives

1. Explain how a chemical equation is written.
2. Explain how to write a balanced equation.
3. Write the steps for writing a balanced chemical equation.

Vocabulary

- | | |
|----------------------|----------------------|
| 1. chemical equation | 4. coefficient |
| 2. stoichiometric | 5. balanced equation |
| 3. reactant | |

Part A Completion

Use the vocabulary words in each sentence, writing either the correct word form or the correct chemical equation. Write each answer on a separate sheet of paper, showing your work.

1. A chemical equation that includes information about the state of the substances involved is called a .
2. The chemical formulas of the reactants are on the , the chemical formulas of the products are on the . The number of atoms of each element must be the same on both sides of the equation. This is called the .
3. A chemical equation that includes information about the state of the substances involved is called a .
4. The number of atoms of each element must be the same on both sides of the equation. This is called the .
5. The number of atoms of each element must be the same on both sides of the equation. This is called the .
6. The number of atoms of each element must be the same on both sides of the equation. This is called the .
7. The number of atoms of each element must be the same on both sides of the equation. This is called the .
8. The number of atoms of each element must be the same on both sides of the equation. This is called the .
9. The number of atoms of each element must be the same on both sides of the equation. This is called the .
10. The number of atoms of each element must be the same on both sides of the equation. This is called the .

1. equation
2. balanced
3. stoichiometric
4. reactant
5. balanced equation
6. coefficient
7. stoichiometric
8. reactant
9. balanced equation
10. coefficient

Section Review 11 1 Describing Chemical Reactions

Answer Key

J Dewey



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