



Unit 1: Chemical Changes

Chemical Changes

Read the passage and answer the following questions.



1. What is the chemical formula for hydrogen peroxide?

Hydrogen peroxide is the chemical formula for the breaking of molecules. In this chemical reaction, the reactants are hydrogen peroxide (H_2O_2) and the products are water (H_2O) and oxygen (O_2). The reaction is exothermic, meaning it releases heat and energy.

2. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product? Is it a catalyst or an inhibitor?

3. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

Hydrogen peroxide is a colorless liquid with a strong, pungent odor. It is a powerful oxidizing agent and is used in many industrial and household applications. In this reaction, it is broken down into water and oxygen. The reaction is exothermic, meaning it releases heat and energy. The products are water and oxygen, which are both colorless and odorless. The reaction is a chemical change because the reactants are broken down into new substances with different chemical properties.

4. According to the law of conservation of mass, if the total mass of the reactants is 100 grams, what is the total mass of the products? Is it 100 grams?



5. What is the chemical formula for water? Is it a liquid, solid, or gas? Is it a reactant or a product?

6. What is the chemical formula for oxygen? Is it a liquid, solid, or gas? Is it a reactant or a product?

7. What is the chemical formula for hydrogen peroxide? Is it a liquid, solid, or gas? Is it a reactant or a product?

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10. According to the law of conservation of mass, if the total mass of the reactants is 100 grams, what is the total mass of the products? Is it 100 grams?

Section 3 Reinforcement Physical And Chemical Changes Answers

John S. Phillips, Cheryl Wistrom



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