

13.1

THE NATURE OF GASES

Section Review

Objectives

1. describe the characteristics of matter having gaseous properties
2. distinguish between a gas and vapour using
3. define the macroscopic and kinetic approaches to understanding gases

Vocabulary

- | | | |
|-----------------|-------------------------|----------------------------------|
| a. condensing | b. vacuum | c. pressure |
| d. condensation | e. atmospheric pressure | f. absolute atmospheric pressure |
| g. properties | h. behaviour | |

Part A Completion

Use the completion format to identify each variable in the sentence, replacing the word(s) indicated below the blank with a word or phrase that best completes the sentence.

Flammable gases describe the _____, expanding characteristics of gases exhibiting the _____ the ability to expand to occupy the largest available free space (note: _____ the tendency to expand relative to _____, _____, and _____ other states of matter). _____ matter the _____ characteristics responsible for the behaviour that _____ pressure exerted. The pressure exerted against a container is determined by the _____ of the gas particles and is related to temperature and is defined as _____ pressure. Absolute pressure is defined as measured for _____ with pressure at _____.

1	_____
2	_____
3	_____
4	_____
5	_____
6	_____
7	_____
8	_____
9	_____
10	_____

Part B True-False

Mark each statement as either **True** or **False**. Indicate the answer by writing **True** or **False** in the space provided.

- | | |
|--|--|
| _____ 11. condensing properties describe | _____ 12. pressure exerted against a container |
| _____ 13. pressure exerted against a container | _____ 14. pressure exerted against a container |

Section 1 The Nature Of Gases Answers

SB Merriam



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