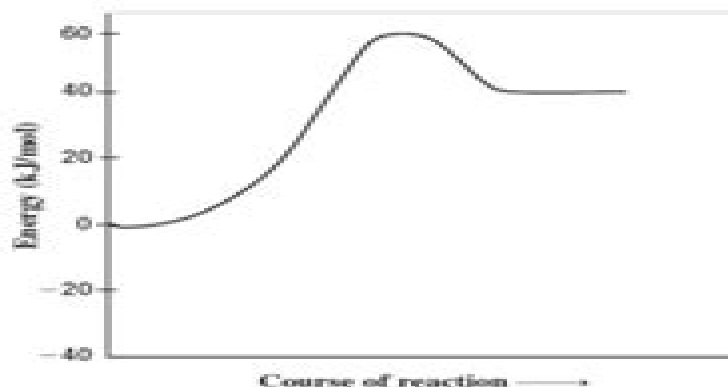


MIXED REVIEW continued

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

4. Answer the following questions using the energy diagram below.



- Is the forward reaction represented by the curve exothermic or endothermic?
- Estimate the magnitude and sign of $\Delta E_{\text{forward}}$.
- Estimate E_a .

A catalyst is added to the reaction, which lowers E_a by about 15 kJ/mol.

- Does the forward reaction rate speed up or slow down?
 - Does the reverse reaction rate speed up or slow down?
 - Does $\Delta E_{\text{forward}}$ change from its value in part b?
5. a. Determine the overall balanced equation for a reaction having the following proposed mechanism:



- Which is the rate-determining step?
- What is the intermediate in the above reaction?

Section 17 Study Guide Answer Key

**Emily Slone McKinney, Sharon Smith
Murray**



Section 17 Study Guide Answer Key:

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
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