Behavior of Gases

increase

Charles's

force

particles

Boyle's	absolute	decrease	larger	temperature
constantly	liquids	pressure	decrease	increased
Gases in Ear	th's atmosphere e	exert 1	0	n everything.
According to the 2.		theory, the particles of a gas are		
3	mo	ving. Every time g	gas particles hit s	omething and
bounce off, th	ey exert a tiny for	ce. Pressure is thi	s amount of	
4	exe	rted per unit of a	rea. Air pressure	at sea level
is 101.3 5				
The amount	of force exerted	by a gas depends	on the 6	
of its container. 7.		law states that if a sample of gas		
is kept at constant 8		decreasing the volume will		
				se the volume, the
pressure will I	10	Accor	ding to the kinet	tic theory, if you do
not change th	e amount of gas	or its temperature	but 11	
the size of the	container, the pa	rticles will strike	the walls more o	ften and the pres-
sure will rise.	When the size of	the container is 1:	2	, the
pressure is sm	aller because the	13	hit t	he walls less often.
		law,		
		, the volume in		
		Charles's measures		
17	of	a gas would beco	me zero at a ten	perature of
-273° C. The t	emperature -273	°C is called 18		zero. All
gases become 19.			cooled to their	
20	P	oints.		

kinetic

size

pressure

boiling

volume

kilopascals

Section 13 Gases Worksheet Answer Key

R Pring

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