

CHEMICAL REACTIONS

Practice Problems

In your notebook, solve the following problems. Use the 3-step problem-solving approach you learned in Chapter I.

SECTION 11.1 DESCRIBING CHEMICAL REACTIONS

- Write the skeleton equation for the reaction between hydrogen and oxygen that produces water.
- Write the skeleton equation for the reaction that produces iron(II) sulfide from iron and sulfur.
- Write the skeleton equation representing the heating of magnesium carbonate to produce solid magnesium oxide and carbon dioxide gas.
- 4. Write a balanced equation for the production of HCl gas from its elements.
- Write a sentence that completely describes the chemical reaction represented by this balanced equation.

 $2HCl(aq) + CaCO_3(s) \rightarrow CO_2(g) + CaCl_2(aq) + H_2O(l)$

Write the word equation for the following equation. Write a sentence fully describing the reaction. Is the equation correctly balanced? Explain.

 $2Ag(s) + S(s) \rightarrow Ag_{s}S(s)$

- Write a balanced equation representing the formation of aqueous sulfuric acid from water and sulfur trioxide gas.
- 8. Write a balanced equation from this word equation.

aqueous silver nitrate + copper metal -- silver metal + aqueous copper nitrate

9. Write a balanced equation for the following word equation.

phosphorus + oxygen → tetraphosphorous decoxide

SECTION 11.2 TYPES OF CHEMICAL REACTIONS

- Write a balanced equation representing the reaction of magnesium with oxygen gas to produce magnesium oxide.
- Write the balanced equation for the reaction that occurs between aluminum and fluorine.
- Write the balanced equation for the production of oxygen gas and potassium chloride from the decomposition of potassium chlorate.
- Write the balanced equation for the reaction between hydrochloric acid and calcium metal. The products are hydrogen gas and calcium chloride.
- Write the balanced equation for the combustion of propane (C₃H₈) to produce carbon dioxide and water vapor.
- Write the balanced equation for the reaction between iron(III) chloride and sodium hydroxide. The products are iron(III) hydroxide and sodium chloride.

Section 11 1 Describing Chemical Reactions Answer Key Workbok

Zhe Xu

Section 11 1 Describing Chemical	Reactions Answer Key Workbok:

Reviewing **Section 11 1 Describing Chemical Reactions Answer Key Workbok**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "Section 11 1 Describing Chemical Reactions Answer Key Workbok," an enthralling opus penned by a very acclaimed wordsmith, readers attempt an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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