Chapter 4

Section 2 A Guided Tour of the Periodic Table

Objectives

- Relate the organization of the periodic table to the arrangement of electrons within an atom.
- Explain why some atoms gain or lose electrons to form ions.
- Determine how many protons, neutrons, and electrons an atom has, given its symbol, atomic number, and mass number.
- Describe how the abundance of isotopes affects an element's average atomic mass.

Chapter menu

Resources

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