



Lewis Structures

- The pair of dots representing a shared pair of electrons in a covalent bond is often replaced by a long dash.
 - example: $\text{H}-\text{H}$ $\text{:}\ddot{\text{F}}-\ddot{\text{F}}\text{:}$
- Structural formula** - indicates the kind, number, and arrangement, and bonds but not the unshared pairs of the atoms in a molecule.
 - example: $\text{F}-\text{F}$ $\text{H}-\text{Cl}$



Section 6 2 Covalent Bonding

S Marginson



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