

SECTION 5.2 ELECTRON ARRANGEMENT IN ATOMS (pages 133–136)

This section shows you how to apply the aufbau principle, the Pauli exclusion principle, and Hund's rule to help you write the electron configurations of elements. It also explains why the electron configurations for some elements differ from those assigned using the aufbau principle.

► Electron Configurations (pages 133–135)

1. The ways in which electrons are arranged around the nuclei of atoms are called _____ electron configurations _____.

Match the name of the rule used to find the electron configurations of atoms with the rule itself.

b _____ 2. aufbau principle

c _____ 3. Pauli exclusion principle

a _____ 4. Hund's rule

- a. When electrons occupy orbitals of equal energy, one electron enters each orbital until all the orbitals contain one electron with parallel spins.
b. Electrons enter orbitals of lowest energy first.
c. An atomic orbital may describe at most two electrons.

5. Look at the aufbau diagram, Figure 5.7 on page 133. Which atomic orbital is of higher energy, a $4f$ or a $5p$ orbital? _____ $4f$

6. Fill in the electron configurations for the elements given in the table. Use the orbital filling diagrams to complete the table.

Element 1	Orbital filling						Electron configuration
	s	2s	2p _x	2p _y	2p _z	3s	
H	☐	☐	☐	☐	☐	☐	1s
He	☐	☐	☐	☐	☐	☐	1s ²
Li	☐	☐	☐	☐	☐	☐	1s ² 2s
C	☐	☐	☐	☐	☐	☐	1s ² 2s ² 2p ²
N	☐	☐	☐	☐	☐	☐	1s ² 2s ² 2p
O	☐	☐	☐	☐	☐	☐	1s ² 2s ² 2p ⁴
F	☐	☐	☐	☐	☐	☐	1s ² 2s ² 2p
Ne	☐	☐	☐	☐	☐	☐	1s ² 2s ² 2p ⁶
Na	☐	☐	☐	☐	☐	☐	1s ² 2s ² 2p ⁶ 3s

Section 4 3 Review Arrangement Of Electrons In Atoms

David Kirk



Section 4 3 Review Arrangement Of Electrons In Atoms:

Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 **Elements of Physical Chemistry** Peter William Atkins, Julio De Paula, 2017 This revision of the introductory textbook of physical chemistry has been designed to broaden its appeal particularly to students with an interest in biological applications *CliffsNotes Chemistry Quick Review, 2nd Edition* Robyn L Ford, Charles Henrickson, Harold D Nathan, 2011-06-28 Inside the Book Elements Atoms Atomic Structure Electron Configurations Chemical Bonding Organic Compounds States of Matter Gases Solutions Acids and Bases Oxidation Reduction Reactions Electrochemistry Equilibrium Thermodynamics Review Questions Resource Center Glossary Why CliffsNotes Go with the name you know and trust Get the information you need fast CliffsNotes Quick Review guides give you a clear concise easy to use review of the basics Introducing each topic defining key terms and carefully walking you through sample problems this guide helps you grasp and understand the important concepts needed to succeed Access 500 additional practice questions at www.cliffsnotes.com/go/quiz/chemistry Master the Basics Fast Complete coverage of core concepts Easy topic by topic organization Access hundreds of practice problems at www.cliffsnotes.com/go/quiz/chemistry Chemical Abstracts, 1927 **Prentice Hall Chemistry**, 2000 **Basic Concepts of Chemistry, Study Guide** Leo J. Malone, 1989-02-21 This Third Edition revised to provide smoother transitions between topics employs a concise yet informal approach to basic chemistry organized to help students employ basic math skills and problem solving strategies Writing style is straightforward and presentation incorporates many concrete analogies to clarify new concepts Includes many illustrative worked examples **General Electric Review** General Electric Company, 1917 **Matter** Prentice-Hall Staff, 1994 **Proceedings of the American Institute of Electrical Engineers**, 1920 Journal of the American Institute of Electrical Engineers American Institute of Electrical Engineers, 1920 Includes preprints of Transactions of the American Institute of Electrical Engineers ISSN 0096 3860 **Advances in Atomic, Molecular, and Optical Physics**, 2006-12-13 Volume 54 of the Advances in Atomic Molecular and Optical Physics Series contains ten contributions covering a diversity of subject areas in atomic molecular and optical physics The article by Regal and Jin reviews the properties of a Fermi degenerate gas of cold potassium atoms in the crossover regime between the Bose Einstein condensation of molecules and the condensation of fermionic atom pairs The transition between the two regions can be probed by varying an external magnetic field Sherson Julsgaard and Polzik explore the manner in which light and atoms can be entangled with applications to quantum information processing and communication They report on the result of recent experiments involving the entanglement of distant objects and quantum memory of light Recent developments in cold Rydberg atom physics are reviewed in the article by Choi Kaufmann Cubel Liebisch Reinhard and Raithel Fascinating experiments are described in which cold highly excited atoms Rydberg atoms and cold plasmas are generated Evidence for a collective excitation of Rydberg matter is also presented Griffiin and Pindzola offer an account of non perturbative quantal

methods for electron atom scattering processes Included in the discussion are the R matrix with pseudo states method and the time dependent close coupling method An extensive review of the R matrix theory of atomic molecular and optical processes is given by Burke Noble and Burke They present a systematic development of the R matrix method and its applications to various processes such as electron atom scattering atomic photoionization electron molecule scattering positron atom scattering and atomic molecular multiphoton processes Electron impact excitation of rare gas atoms from both their ground and metastable states is discussed in the article by Boffard Jung Anderson and Lin Excitation cross sections measured by the optical method are reviewed with emphasis on the physical interpretation in terms of electronic structure of the target atoms Ozier and Moazzen Ahmadi explore internal rotation of symmetric top molecules Developments of new experimental methods based on high resolution torsional vibrational and molecular beam spectroscopy allow accurate determination of internal barriers for these symmetric molecules The subject of attosecond and angstrom science is reviewed by Niikura and Corkum The underlying physical mechanisms allowing one to generate attosecond radiation pulses are described and the technology needed for the preparation of such pulses is discussed LeGou t Bretenaker and Longer describe how rare earth ions embedded in crystals can be used for processing optically carried broadband radio frequency signals Methods for reaching tens of gigahertz instantaneous bandwidth with submegahertz resolution using such devices are analyzed in detail and demonstrated experimentally Finally in the article by Illing Gauthier and Roy it is shown that small perturbations applied to optical systems can be used to suppress or control optical chaos spatio temporal dynamics and patterns Applications of these techniques to communications laser stabilization and improving the sensitivity of low light optical switches are explored International experts Comprehensive articles New developments

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